

JC20 Rec'd PCT/PTO 10 JUN 2005.

(AW 08/03)

Substitute for Form 1449A/PTO		<i>Complete if Known</i>	
		107538341	
		Assigned	
INFORMATION DISCLOSURE STATEMENT BY APPLICANT <i>(Use as many sheets as necessary)</i>			
SHEET 1 of 2		Attorney Docket No. JMYS-130US	

INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(Use as many sheets as necessary)

SHEET 1 of 2

Attorney Docket No. JMYS-130US

U.S. PATENT DOCUMENTS

FOREIGN PATENT DOCUMENTS

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Examiner Initials*	Cite No. ¹	Foreign Patent Document	Publication Date (MM-DD-YYYY)	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines, Where Relevant Passages or Relevant Figures Appear	T ⁶
		Country Code ³ - Number ⁴ - Kind Code ⁵ (if known)				
/FT/	2	DE-29 47 978-A1	06-12-1980	Daicel Chemical Industries, Ltd.		<input type="checkbox"/>
/FT/	3	EP-0 943 641-A1	09-22-1999	Daicel Chemical Industries, Ltd.		<input type="checkbox"/>
/FT/	4	EP-0 710 685-B1	01-21-2004	Nippon Shokubai Co., Ltd.		<input type="checkbox"/>
/FT/	5	JP-4-257545	09-11-1992	Mitsubishi Rayon Co. Ltd.		<input type="checkbox"/>
/FT/	6	WO-03/018662-A1	03-06-2003	Johnson Matthey PLC		<input type="checkbox"/>

Examiner Signature	/Fred Teskin/	Date Considered	12/30/2007
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*EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to Applicant.

'Applicant's unique citation designation number (optional):

²See Kind Codes of USPTO Patent Documents at www.uspto.gov or MPEP 901.04.

³Enter Office that issued the document, by the two-letter code (WIPO Standard St.3).

***For Japanese patent documents, the indication of the year of the reign of the Emperor must precede the serial number of the patent document.**

⁵Kind of document by the appropriate symbols as indicated on the document under WIPO Standard ST.16 if possible.

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The collection of information is required by 37 CFR 1.97 and 1.98. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 2 hours to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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Substitute for Form 1449B/PTO		Complete if Known-	
INFORMATION DISCLOSURE STATEMENT BY APPLICANT <i>(Use as many sheets as necessary)</i>		Application Number	To Be Assigned <i>100-538341</i>
		Filing Date	Herewith
		First Named Inventor	Martin Graham Partridge
		Art Unit	
		Examiner Name	
SHEET 2 of 2		Attorney Docket No.	JMYS-130US

NON-PATENT LITERATURE DOCUMENTS			
Examiner Initials*	Cite No. ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published	T ²
/FT/	7	Brendan J. O'Keefe, Marc A. Hillmyer and William B. Tolman, "Polymerization of lactide and related cyclic esters by discrete metal complexes," <i>J. Chem. Soc., Dalton Trans.</i> , 2001, pp. 2215-2224.	<input type="checkbox"/>
/FT/	8	Youngjo Kim and John G. Verkade, "Novel Titanatranes with Different Ring Sizes: Syntheses, Structures, and Lactide Polymerization Catalytic Capabilities," <i>Organometallics</i> , Vol. 21, No. 12, 2002, pp. 2395-2399.	<input type="checkbox"/>
/FT/	9	Björn Antelmann, Malcolm H. Chisholm, Suri S. Iyer, John C. Huffman, Diana Navarro-Llobet, and Marty Pagel, "Molecular Design of Single Site Catalyst Precursors for the Ring-Opening Polymerization of Cyclic Ethers and Esters. 2. Can Ring-Opening Polymerization of Propylene Oxide Occur by a Cis-Migratory Mechanism?" <i>Macromolecules</i> , Vol. 34, No. 10, 2001, pp. 3159-3175.	<input type="checkbox"/>
/FT/	10	Daisuke Takeuchi, Takeshi Nakamura, and Takuzo Aida, "Bulky Titanium Bis(phenolate) Complexes as Novel Initiators for Living Anionic Polymerization of ϵ -Caprolactone," <i>Macromolecules</i> , Vol. 33, No. 3, 2000, pp. 725-729.	<input type="checkbox"/>
/FT/	11	Huei-Ling Chen, Bao-tsan Ko, Bor-Hunn Huang, and Chu-chieh Lin, "Reactions of 2,2'-Methylenebis(4-chloro-6-isopropyl-3-methylphenol) with Trimethylaluminum: Highly Efficient Catalysts for the Ring-Opening Polymerization of Lactones," <i>Organometallics</i> , Vol. 20, No. 24, 2001, pp. 5076-5083.	<input type="checkbox"/>
/FT/	12	Malcolm H. Chisholm, Nancy W. Eilerts, John C. Huffman, Suri S. Iyer, Martha Pacold, and Khamphhee Phomphrai, "Molecular Design of Single-site Metal Alkoxide Catalyst Precursors for Ring-Opening Polymerization Reactions Leading to Polyoxygenates. 1. Polylactide Formation by Achiral and Chiral Magnesium and Zinc Alkoxides, (η^3 -L) MOR, Where L = Trispyrazolyl- and Trisindazolylborate Ligands," <i>J. Am. Chem. Soc.</i> , Vol. 122, No. 48, 2000, pp. 11845-11854.	<input type="checkbox"/>
			<input type="checkbox"/>

Examiner Signature	/Fred Teskin/	Date Considered	12/30/2007
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